Calculating the Feed Oxygen Demand (FOD) of Aquafeeds

Claude E. Boyd

ABSTRACT

The biological oxygen demand of aquafeeds will be called the feed oxygen demand (FOD). It is the amount of dissolved oxygen needed to oxidize organic carbon and ammoniacal nitrogen applied to culture systems in feeds and not recovered in biomass of the culture species at harvest. The following equation can be used to estimate FOD:

$$FOD = [C_f - (FCE \times C_a)] \, 2.67 + [N_f - (FCE \times N_a)] \, 4.57$$

where, C_f, N_f, C_a, N_a = decimal fractions (% ÷ 100) of carbon and nitrogen in feed and culture animals, respectively; FCE = feed conversion efficiency (1 ÷ FCR). The FOD value will vary with species, feed quality, feeding practices, and feed conversion efficiency, but calculations suggest that FOD usually will be 1.2 to 1.3 kg O_2/kg feed.

The biological oxygen demand or BOD is widely recognized as an important water quality variable. It is used mainly as an index of the pollution strength of municipal, industrial, and agricultural waste waters (Eaton *et al.* 2005). The BOD of an effluent is a measure of the amount of oxygen needed by microorganisms in oxidizing organic carbon to carbon dioxide and ammoniacal nitrogen to nitrate (Sawyer and McCarty 1967). Oxygen required to oxidize organic carbon is referred to as the carbonaceous biological oxygen demand (CBOD), and oxygen needed to oxidize ammoniacal nitrogen is called the nitrogenous oxygen demand (NOD). The two fractions usually are combined and reported as the BOD. The 5-day BOD (BOD_5) is most commonly determined, and for typical wastewater and aquaculture pond water, the BOD_5 concentration is about two-thirds of the quantity of oxygen needed over a longer period of time to completely oxidize the organic carbon and ammoniacal nitrogen (Boyd and Gross 1999). The amount of oxygen needed to satisfy the BOD of an effluent completely is known as the ultimate BOD (BOD_u). For example, suppose that the BOD_u concentration in an effluent averages 200 mg/L (200 g/m^3) and 250 m^3 of this effluent are discharged into a lake each day. The effluent will impose an oxygen demand of 50 kg/day (200 g BOD/m^3 x 250 m^3/day) on the lake.

Dissolved oxygen concentration is likely the single most important water quality variable in aquaculture (Boyd and Tucker 1998). An adequate concentration of dissolved oxygen is needed to avoid stress and to assure good growth of the culture

Department of Fisheries and Allied Aquacultures Auburn University, Alabama 36849 USA

species. Moreover, sufficient dissolved oxygen is needed in aquaculture systems to oxidize wastes and prevent them from accumulating to levels harmful to fish, shrimp, and other culture species. The primary wastes in aquaculture systems are organic carbon and ammoniacal nitrogen resulting from the fraction of aquafeeds not converted to biomass of the culture species and harvested. The purpose of this report is to present a method for calculating the oxygen demand of aquafeeds.

The Fate of Aquafeeds

The fate of aquafeeds in ponds is illustrated in Fig. 1. The amount of feed eaten by the culture animals varies with species and feeding practices. When care is exercised to avoid overfeeding, some fish species consume 95 % or more of feed pellets. Shrimp, however, nibble their food and seldom consume more than 85 % of feed pellets. Aquafeeds are comprised of high quality ingredients, and up to 85 or 90 % of the eaten feed will be absorbed by the culture species. The uneaten feed and feces will be degraded by microorganisms to carbon dioxide, ammonia, and other inorganic substances. Some of the feed components absorbed by the culture species will be converted to biomass and removed from the pond at harvest. However, the majority of the absorbed nutrients will be oxidized to carbon dioxide through

Fig. 1   Fate of Aquafeeds in ponds.
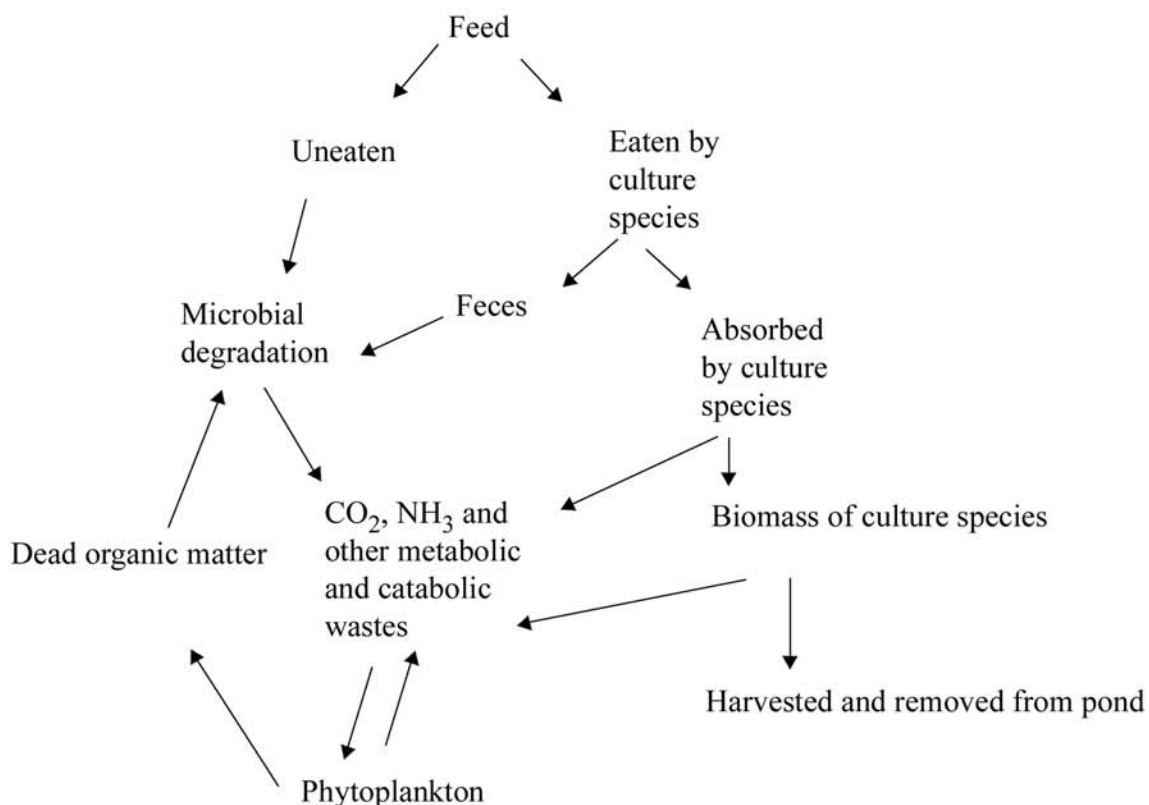
respiration or transformed to ammonia and other metabolic and catabolic wastes and excreted. Usually, no more than 8 to 10 % of the organic carbon and 20 to 30 % of the nitrogen in aquafeeds will be harvested in biomass of the culture species (Boyd *et al.* 2007). Feed nitrogen not recovered in the culture species will be mineralized to ammoniacal nitrogen and oxidized to nitrate by nitrifying bacteria.

Carbon dioxide, inorganic nitrogen, and other nutrients released into pond water following oxidation and mineralization of feed ingredients by the culture species and microorganisms stimulate phytoplankton productivity. In the long run, the amount of oxygen produced in phytoplankton photosynthesis will be equivalent to the amount of oxygen used in respiration by phytoplankton and by microorganisms degrading dead phytoplankton (Fig. 2). Thus, the stimulation of phytoplankton productivity by nutrients from feeding wastes should not be considered as part of the oxygen demand of feed.

Fig.2   As ecological processes, photosynthesis and respiration are equal but opposite reactions. The amounts of organic carbon and oxygen produced in photosynthesis are equal to the quantities of inorganic carbon released and oxygen used in respiration.

Photosynthesis

$$6CO_2 + 6H_2O + Energy + Nutrients \longrightarrow C_6H_{12}O_6 + 6O_2$$

Phytoplankton

Organic matter

Respiration
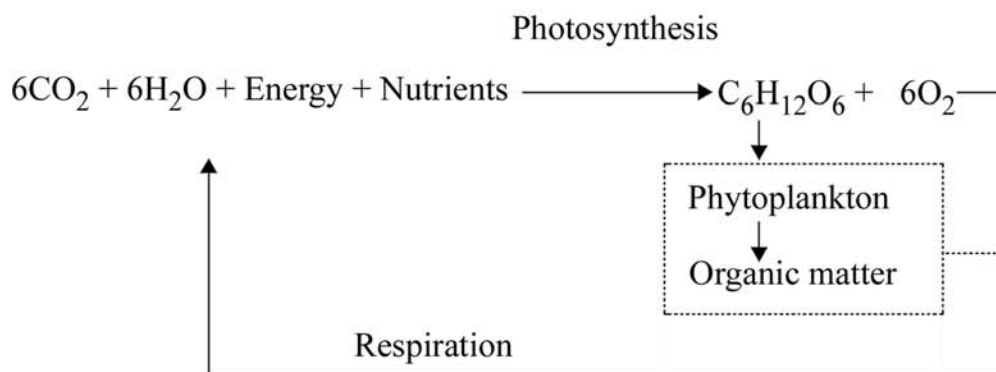
Oxygen Demand of Aquafeeds

The oxygen demand of aquafeeds will be called the feed oxygen demand or FOD. It will be defined as the amount of oxygen required to oxidize all organic carbon and ammoniacal nitrogen applied to ponds and other aquaculture systems in feed and not recovered in biomass of the culture species at harvest. The summary equation for oxidation of organic carbon by the culture species or by microorganisms is:

$$Organic\ C + O_2 \rightarrow CO_2. \qquad (1)$$

The ratio of molecular weights of oxygen to carbon (O_2:C) in the above equation is 32:12 or 2.67:1. The summary equation for oxidation of ammoniacal nitrogen to nitrate nitrogen by nitrifying bacteria is:

$$NH_4^+ + 2O_2 \rightarrow NO_3^- + 2H^+ + H_2O. \qquad (2)$$
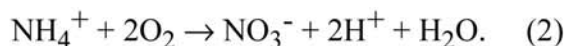
The ratio of molecular weights of oxygen to ammoniacal nitrogen is $2O_2$:N (64:14 or 4.57:1).

Based upon the stoichiometry of Equations 1 and 2, the simple expression for FOD is:

$$FOD = (Feed\ C - Harvest\ biomass\ C)\ 2.67 + (Feed\ N - Harvest\ biomass\ N)\ 4.57. \quad (3)$$

Of course, the value of FOD will vary with the percentages of carbon and nitrogen in aquafeeds and culture species, and with the efficiency with which feed is converted to biomass of the culture species. The calculation of FOD will be illustrated for an example from shrimp aquaculture in which the feed conversion ratio (FCR) was 1.8, the feed contained 6 % nitrogen and 45 % carbon, and the shrimp were 2.75 % nitrogen and 11 % carbon (Boyd and Teichert -Coddington 1995). Each kilogram of feed contained 0.45 kg carbon, 0.06 kg nitrogen and produced 0.556 kg shrimp containing 0.015 kg nitrogen and 0.061 kg carbon. The FOD may be estimated as follows:

$$FOD = (0.45 - 0.061)\ 2.67 + (0.06 - 0.015)\ 4.57 = 1.039 + 0.206 = 1.245\ kg\ O_2/kg\ feed.$$

Notice that the carbonaceous fraction of the FOD was about five times greater than the nitrogenous fraction of the FOD in this example.

The calculations illustrated above may be combined into the following general equation:

$$FOD = [C_f - (FCE \times C_a)]\ 2.67 + [N_f - (FCE \times N_a)]\ 4.57 \quad (4)$$

where, C_f, N_f, C_a, N_a = decimal fractions (% ÷ 100) of carbon and nitrogen in feed and culture animals, respectively; FCE = feed conversion efficiency (1 ÷ FCR).

Aquafeeds usually contain 45 to 50 % organic carbon and 4.5 to 7 % nitrogen (28 to 44 % crude protein). Carbon and nitrogen concentrations vary with species, but for the present discussion, it will be assumed that culture species are 11 % organic carbon and 2.75 % nitrogen (see shrimp example above). For feeds containing 45 % organic carbon, one with 4.5 % nitrogen would have FOD = 1.176 kg O_2/kg feed, while one with 7 % nitrogen would have FOD = 1.290 kg O_2/kg feed. Feeds for coldwater fish, e.g., trout and salmon, have high percentages of organic carbon and nitrogen, but FCR usually is 1.1 to 1.2 (Boyd et al. 2007). At FCR of 1.15, a feed with 50 % C and 7 % N would have FOD = 1.289 kg O_2/kg feed.

The FCR varies with species, feed quality, and method of feeding, and values typically are between 1.5 and 2.5. Using the shrimp example from above, FOD will vary from 1.199 kg O_2/kg feed at FCR = 1.5 to 1.308 kg O_2/kg feed at FCR = 2.5 (Table 1). It is important to note that decreasing (improving) the FCR will reduce the FOD input per kilogram of shrimp production. For example, at a FCR of 2.2, the FOD input is 2.82 kg O_2/kg shrimp, while at a FCR of 1.6, the FOD input is reduced to 1.94 kg O_2/kg shrimp.

The data in Table 1 suggest that the FOD in shrimp culture will be around 1.2

to 1.3 kg O_2/kg feed applied to ponds, and the FOD for culture of most species of fish likely will be similar. Equation 4 can be used to obtain a more exact estimate of FOD from the anticipated FCR and data on the carbon and nitrogen composition of the feed and culture species. The FOD likely is similar across species and types of feed. Thus, the most important action that can be taken to reduce the oxygen demand of feed in aquaculture systems is to improve the efficiency of conversion of feed to culture species.

Table 1.   Feed conversion ratio (FCR), feed conversion efficiency (FCE), feed oxygen demand (FOD), and FOD input for culture of shrimp.

FCR	FCE	FOD (kg O_2/kg feed)	FOD input (kg O_2/kg shrimp)
1.5	0.667	1.199	1.80
1.6	0.625	1.214	1.94
1.7	0.588	1.229	2.09
1.8	0.556	1.245	2.24
1.9	0.526	1.257	2.39
2.0	0.500	1.265	2.53
2.1	0.476	1.278	2.68
2.2	0.455	1.283	2.82
2.3	0.435	1.292	2.97
2.4	0.417	1.303	3.13
2.5	0.400	1.308	3.27

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